

SYNTHESIS OF 7- (OR 6-)HYDROXY DERIVATIVES OF 3,3-DIMETHYL- 3,4-DIHYDROISOQUINOLINE

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When treated with cyanoacetic ester in conc. H₂SO₄ 1-(4-hydroxy-3-methoxyphenyl)- and 1-(3-hydroxy-4-methoxyphenyl)-2-methylpropan-1-ol form ethyl 7-hydroxy-6-methoxy-3,3-dimethyl(or 6-hydroxy-7-methoxy- 3,3-dimethyl)-1,2,3,4-tetrahydroisoquinolydene-1-acetate.

Keywords: vanillin, 3,3-dimethyldehydroisalsoline, 3,3-dimethyldehydroisalsoline, isovanillin.

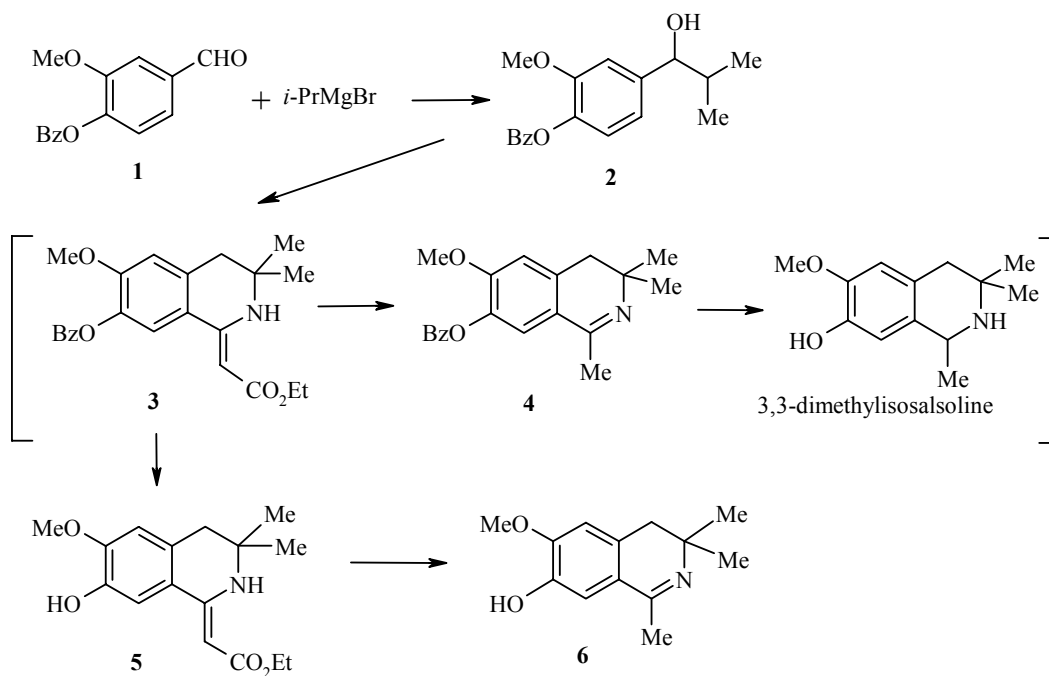
6(or 7-)-Hydroxy-7-(or-6-)methoxy derivatives of isoquinolines hydrogenated in the heterocyclic fragment are quite widely distributed in nature. These include e.g. salsoline, isosalsoline, N-methylisalsoline, lophocerine, corypaline, O-methylanhalonidine, pseudolaudanine, and many others [1-8]. At the same time, 3,3-dimethyl analogs of these compounds have hardly been studied. It can be recalled that 3,3-dimethyl-3,4-dihydroisoquinoline N-oxide [9] is a powerful agent used in septic and traumatic shock and in a recent US patent [10] relating to preparation of 1-substituted 3,3,4,4-tetramethyl-3,4-dihydroisoquinolines. It is clear that reduction of 3,4-dihydroisoquinolines to 1,2,3,4-tetrahydro derivatives is a routine problem.

The simplest method for preparing 3,3-dialkyl-3,4-dihydroisoquinolines is the reaction of dimethylbenzylcarbinol with nitriles [11] or a three-component reaction of an activated arene, isobutyraldehyde, and nitriles in conc. H₂SO₄ [12]. It is known that free phenols are readily sulfonated in conc. H₂SO₄ hence the simplest method for synthesis of 7-hydroxy-6-methoxy-1,3,3-trimethyl-1,2,3,4-tetrahydroisoquinolines is the reaction of carbinol **2** (prepared from the 4-benzyloxy-3-methoxybenzaldehyde (**1**) substituted at the phenol oxygen atom using isopropyl magnesium bromide) and cyanoacetic ester with subsequent hydrolysis of the ethyl 7-benzyloxy-6-methoxy-3,3-dimethyl-1,2,3,4-tetrahydroisoquinolydene-1-acetate (**3**) to 7-benzyloxy-6-methoxy-1,3,3-trimethyl-3,4-dihydroisoquinoline (**4**) and removal of the benzyl protection by hydrogenation with simultaneous reduction to the tetrahydro derivative.

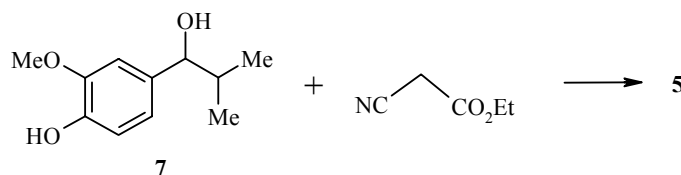
It was found that, the cyclization reaction conditions to compound **3** were accompanied by removal of the benzyl group and immediate formation of the 7-hydroxy derivative **5** which was readily transformed to the 3,3-dimethyldehydroisalsoline **6**. A small amount of compound **3** was also separated in which the benzyl group was retained.

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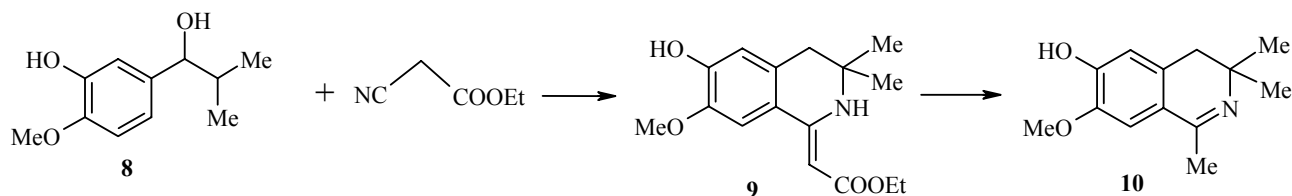
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Separation of compound 5 from the reaction product led us to study the possibility of its synthesis from carbinol 7 prepared directly from vanillin and this gave an almost 2.5 times increase in the yield of the target compound 5.



A similar reaction with carbinol 8 prepared from isovanillin gave compound 9 in 24% yield. This is due to the difficulty of crystallizing ester 9 since chromatographic data on the neutralized reaction product shows the yields of compounds 5 and 9 from the corresponding carbinols to be around 70%. Ester 9 is readily converted to 3,3-dimethyldehydroisosalsole 10 *via* hydrolysis with 10% sulfuric acid.



Hence we have discovered routes to preparation of 7(or 6)-hydroxy-3,4-dihydroisoquinolines which open up broad possibilities for synthesizing analogs of known alkaloids modified at the ring 3 position.

EXPERIMENTAL

IR spectra were recorded on a UR-20 instrument using vaseline oil. ¹H NMR spectra were registered on a MercuryPlus-300 spectrometer (300 MHz) using DMSO-d₆ and HMDS internal standard. Mass spectra were

obtained on an Agilent GC 6890N MSD 5975B spectrometer (EI, 70 eV). Elemental analysis was performed on a CHNS-932 LECO Corporation analyzer. Monitoring of the reaction course and purity of the compounds obtained was carried out by TLC on Silufol plates in the system chloroform–acetone (9:1) and revealed using a 5% solution of chloranil in toluene.

Carbinols **2**, **7** and **8** were prepared *via* organomagnesium synthesis from isopropylmagnesium bromide and benzylvanillin, vanillin, and isovanillin respectively as a mixture with the corresponding styrene (~ 3:2 by LC) and were used without additional purification [13].

Synthesis of Compounds 3, 5, 6, 9, and 10 (General Method). A mixture of 1-(4-hydroxy-3-methoxyphenyl)-2-methylpropanol (**7**) (for preparing compounds **5** and **6**) or 1-(3-hydroxy-4-methoxyphenyl)-2-methylpropanol (**8**) (for compounds **9** and **10**) (1.95 g, 0.01 mol) and cyanoacetic ester (1.13 g, 0.01 mol) was added dropwise with vigorous stirring to 98% H₂SO₄ (8 ml) cooled to 5–10°C. The product was stirred for 30 min at room temperature. The reaction mixture was poured into water and neutralized using sodium carbonate to pH 7–8, extracted with chloroform (3×50 ml), and the combined organic layer was washed with water and dried over magnesium sulphate. Chloroform was distilled off and the residue was crystallized from the appropriate solvent. Compound **3** was separated by fractional crystallization of the mixture of esters **3** and **5** from methanol. The yield of compound **5** by this method was 22%.

Compounds **6** and **10** were prepared by refluxing compounds **5** and **9** in 10% H₂SO₄ for 3 h, cooling to room temperature, and subsequent separation as reported above.

Ethyl 7-benzyloxy-6-methoxy-3,3-dimethyl-1,2,3,4-tetrahydroisoquinolylidene-1-acetate (3). Mp 207–209°C (methanol); yield 8%. IR spectrum, ν , cm⁻¹: 3360 (NH), 1630 (sh, C=O); 1620 (C=N). ¹H NMR spectrum, δ , ppm (*J*, Hz): 1.09 (3H, t, *J* = 7.2, OCH₂CH₃); 1.17 (6H, s, 3,3-(CH₃)₂); 2.70 (2H, s, H-4); 3.87 (3H, s, 6-OCH₃); 3.92 (2H, q, *J* = 7.2, OCH₂CH₃); 4.24 (2H, s, OCH₂C₆H₅); 4.54 (1H, s, CH=CH–); 6.80 (1H, s, H-5); 8.40 (1H, s, H-8). Mass spectrum, *m/z* (*I*_{rel.}, %): 381 [M]⁺ (60), 352 (10), 336 (13), 290 (35), 262 (20), 230 (14), 178 (19), 165 (20).

Ethyl 7-Hydroxy-6-methoxy-3,3-dimethyl-1,2,3,4-tetrahydroisoquinolylidene-1-acetate (5). Mp 148–151°C (ethanol); yield 56% (carbinol **7**). IR spectrum, ν , cm⁻¹: 3460, 3270, 1640, 1602, 1574, 1512. ¹H NMR spectrum, δ , ppm (*J*, Hz): 1.25 (6H, s, 3,3-(CH₃)₂); 1.30 (3H, m, OCH₂CH₃); 2.72 (2H, s, H-4); 3.91 (3H, s, 6-OCH₃); 4.12 (2H, q, *J* = 7.2, OCH₂CH₃); 5.00 (1H, s, –CH=C–); 5.50 (1H, br. s, 7-OH); 6.58 (1H, s, H-5); 7.21 (1H, s, H-8); 8.90 (1H, br. s, NH). Mass spectrum, *m/z* (*I*_{rel.}, %): 291 [M]⁺ (52), 276 [M-CH₃]⁺ (20), 246 [M-OC₂H₅]⁺ (20), 230 [M-OOC₂H₅]⁺ (100), 219 (23), 204 (27). Found, %: C 66.18; H 7.21; N 4.85. C₁₆H₂₁NO₄. Calculated, %: C 65.96; H 7.27; N 4.81.

7-Hydroxy-6-methoxy-1,3,3-trimethyl-3,4-dihydroisoquinoline (6). Mp 194–204°C (a mixture of ethyl acetate and hexane); yield 53% (29% based on the starting carbinol **7**). IR spectrum, ν , cm⁻¹: 1632, 1612, 1560, 1512. ¹H NMR spectrum, δ , ppm: 1.19 (6H, s, 3,3-(CH₃)₂); 2.29 (3H, s, 1-CH₃); 2.60 (2H, s, H-4); 3.91 (3H, s, 6-OCH₃); 6.61 (1H, s, H-5); 7.07 (1H, s, H-8). Mass spectrum, *m/z* (*I*_{rel.}, %): 219 [M]⁺ (91), 204 [M-CH₃]⁺ (100), 189 (20), 177 (41), 162 (14). Found, %: C 71.29; H 7.76; N 6.44. C₁₃H₁₇NO₂. Calculated, %: C 71.21; H 7.81; N 6.39.

Ethyl 6-Hydroxy-7-methoxy-3,3-dimethyl-1,2,3,4-tetrahydroisoquinolylidene-1-acetate (9). Mp 135–137°C (ethanol); yield 24%. IR spectrum, ν , cm⁻¹: 3280, 3000, 1636, 1592, 1516. ¹H NMR spectrum, δ , ppm (*J*, Hz): 1.25 (6H, s, 3,3-(CH₃)₂); 1.28 (3H, m, OCH₂CH₃); 2.71 (2H, s, H-4); 3.90 (3H, s, 7-OCH₃); 4.15 (2H, q, *J* = 7.2, OCH₂CH₃); 4.99 (1H, s, –CH=C–); 5.86 (1H, s, 6-OH); 6.68 (1H, s, H-5); 7.11 (1H, s, H-8); 8.93 (1H, br. s, NH). Mass spectrum, *m/z* (*I*_{rel.}, %): 291 [M]⁺ (31), 276 [M-CH₃]⁺ (20), 246 [M-OC₂H₅]⁺ (17), 230 [M-OOC₂H₅]⁺ (100), 219 (15), 204 (40). Found, %: C 65.83; H 7.33; N 4.86. C₁₆H₂₁NO₄. Calculated, %: C 65.96; H 7.27; N 4.81.

6-Hydroxy-7-methoxy-1,3,3-trimethyl-3,4-dihydroisoquinoline (10). Mp 195–196°C (mixture of ethyl acetate and hexane); yield 20% (based on starting carbinol **8**). IR spectrum, ν , cm⁻¹: 3200, 1615, 1585, 1512. ¹H NMR spectrum, δ , ppm: 1.30 (6H, s, 3,3-(CH₃)₂); 2.45 (3H, s, 1-CH₃); 2.67 (2H, s, H-4); 3.79 (3H, s,

7-OCH₃); 5.50 (1H, br. s, 6-OH); 6.44 (1H, s, H-5); 6.85 (1H, s, H-8). Mass spectrum, *m/z* (*I*_{rel}, %): 219 [M]⁺ (100), 204 [M-CH₃]⁺ (85), 189 (34), 177 (53), 162 (15). Found, %: C 71.34; H 7.75; N 6.43. C₁₃H₁₇NO₂. Calculated, %: C 71.21; H 7.81; N 6.39.

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REFERENCES

1. A. P. Orekhov, *Chemistry of Alkaloids* [in Russian], Moscow (1955), p. 859.
2. W. J. Gensler, in: R. Elderfield (editor), *Heterocyclic Compounds* [Russian translation], vol. 4, Inostr. Lit., Moscow (1955), p. 264.
3. W. M. Whaley and T. R. Govindachari in: *Organic Reactions* [Russian translation], vol 6, Inostr. Lit., Moscow (1953), p. 98.
4. W. J. Gensler in: *Organic Reactions* [Russian translation], vol. 6, Inostr. Lit, Moscow (1953), p. 218.
5. P. A. Klare in: D. Barton and W. D. Ollis (editors), *Comprehensive Organic Chemistry*, [Russian translation], vol. 8, Khimiya, Moscow (1985), p. 255.
6. M. Shamma, *The Isoquinoline Alkaloids*, Academic Press, New York (1972).
7. M. Shamma and J. L. Moniot, *Isoquinoline Alkaloids Research 1972-1977*, Plenum Press, New York, London (1978), 426 p.
8. A. A. Semenov, *Outline of the Chemistry of Natural Compounds* [in Russian], Nauka, Novosibirsk (2000), 664 p.
9. T. J. N. Watson, *J. Org. Chem.*, **63**, 406 (1998).
10. Mikio Ogawa, Yoshikazu Takaoka, and Akira Ohhata, US Pat. 6956033; www.freepatentsonline.com/6956033.html
11. V. S. Shklyaev, B. B. Aleksandrov, G. I. Legotkina, M. S. Gavrilov, M. I. Vakhrin, and A. G. Mikhailovskii, *Khim. Geterotsykl. Soedin.*, 1560 (1983). [*Chem. Heterocycl. Comp.*, **19**, 1242 (1983)].
12. Yu. V. Shklyaev and Yu. V. Nifontov, *Izv. Akad. Nauk, Ser. Khim.*, 780 (2002).
13. Yu. S. Rozhkova, *Dissertation Cand. Chem. Sci.*, Perm (2006).